

MRS Fall Meeting

In-situ transfer of monolayer graphene fluoride flakes and study by scanning tunneling microscopy

S.W. Schmucker\*, J.W. Wood, J.W. Lyding

The production and characterization of graphene fluoride is of interest due to its retention of the layered structure of graphite and the introduction of a large ( $>3\text{eV}$ ) band gap. Interest in graphene fluoride as an electronic material is extended by the possible reduction of multi-layer graphene fluoride to graphene[1,2]. In this work we demonstrate the exfoliation and characterization of predominantly monolayer graphene fluoride ( $\text{CF}_n$ ) flakes by an in-situ mechanical exfoliation process onto the Si(100) 2x1:H surface and investigate these features by scanning tunneling microscopy (STM). By a dry-contact transfer (DCT) process [3], monolayer graphene fluoride islands are transferred to the passivated silicon surface with negligible substrate contamination, as seen in earlier examples of graphene exfoliation [4]. This ultrahigh vacuum (UHV) compatible transfer enables UHV-STM imaging and electronic characterization of monolayer graphene fluoride islands (12 – 38 nm lateral dimension). The resulting topographic and spectroscopic data suggest local variations in fluorine coverage, which is manifested in variable topographic height, ranging from  $5.3\text{\AA}$  to  $7.3\text{\AA}$  (mean =  $6.42\text{\AA}$ ), and also in the small band gap measured in these islands when probed by scanning tunneling spectroscopy. However, the theoretically anticipated large gap ( $\sim 3\text{ eV}$ ) of graphene fluoride is demonstrated when weakly bound ( $\text{CF}_n$ ) islands are removed from the substrate to the STM tip. Subsequent transport through captured flakes exhibits band gaps of  $2.8 - 3.2\text{ eV}$ .

For ex-situ investigation of transferred few-layer graphene fluoride, graphite fluoride powder is dispersed in N-Methylpyrrolidone (NMP), deposited onto silicon dioxide, and characterized by optical microscopy, atomic force microscopy (AFM), scanning electron microscopy (SEM) and energy-dispersive x-ray spectroscopy (EDX). We thereby verify the transfer of few-layer ( $\text{CF}_n$ ) from NMP and the decomposition and chemical modification of ( $\text{CF}_n$ ) under electron bombardment.

<sup>1</sup>N. Kumagai, et al., J. Appl. Electro. **25**, 869 (1995)

<sup>2</sup>S.-H Cheng, et al., Phys. Rev. B **81**, 205435 (2010)

<sup>3</sup>P.M. Albrecht and J.W. Lyding, Appl. Phys. Lett. **83**, 5029 (2003).

<sup>4</sup>K.A. Ritter and J.W. Lyding, Nat. Mater. **8**, 235 (2009).